

PLATONOVA, V.G.

Economic effectiveness of seed disinfection. Zashch. rast. ot
vred. 1 bol. 8 no.2:10 F '63. (MIRA 16:7)

1. Vsesoyuznyy institut zashchity rasteniy.
(Seeds--Disinfection)

GLEBOV, M.A., kand. sel'skokhoz. nauk; PLATONOVA, V.G., uchenyy agronom

Economic effectiveness of corn protection. Zashch. rast. ot
vred. i bol. 8 no.4:13 Ap '63. (MIRA 16:10)

1. Vsesoyuznyy institut zashchity rasteniy.
(Corn (Maize)---Diseases and pests)
(Spraying and dusting in agriculture)

KOMAROV, V.A.; PLATONOVA, V.I.; RODIMENKOVA, N.A.; KHARITONOV, N.P.;
KHUOBIN, Yu.I.

Effect of alcohol structure and solvent composition on the
kinetics of the alkaline solvolysis of trialkylsilanes.
Zhur. fiz. khim. 38 no.9:2139-2144 S '64. (MIRA 17:12)

1. Institut khimii silikatov imeni Grebenshchikova AN SSSR,
Leningrad.

BABENKO, L.V.; PLATONOVA, V.F.

Diapause in the larvae of *Ixodes ricinus* L. and *Ixodes persulcatus* P.Sch. (Parasitiformes, Ixodidae). Report No.1: Experimental data on the effect of a photoperiod on hungry and satiated larvae. Med. paraz. i paraz. bol. 34 no.1:69-73 Ja-F '65.

(MIRA 18:8)

PLATONOVA, V.A., kand.med.nauk

Pathogenesis of mediastinal subcutaneous emphysema and spontaneous pneumothorax in military tuberculosis in a 7-year-old child. Vop. okh. mat. i det. 6 no.10:83-87 0 '61. (ISSN 14:11)

1. Iz kafedry gospital'noy pediatrii (zav. - kand.med.nauk V.P. Sitnikova) Voronezhskogo meditsinskogo instituta (dir. - prof. N.I.Odnoralov), Oblastnoy detskoy bol'nitsy (glavnyy vrach I.V. Ivanova) Oblastnogo protivotuberkuleznogo dispansera (glavnyy vrach N.S.Pokhvisneva).

(TUBERCULOSIS) (PNEUMOMEDIASTINUM) (PNEUMOTHORAX)

PLATONOVA, V.A., kand.med. nauk

Segmental lung lesion in children with chronic pneumonia.
Pediatriia 4 no.7:21-24 J1'63 (MIRA 16:12)

1. Iz kafedry gospital'noy pediatrii (zav. - kand.med. nauk V.P.Sitnikova) i kafedry rentgenologii s meditsinskoy radiologiyey (zav. - dotsent M.M. Mikhaylov) Voronezhskogo meditsinskogo instituta.

MIKHAYLOV, N.V.; PLATONOV, V.A.

Use of the vapor condensation method for heating viscose
solutions. Khim. volok. no.2:40-43 '59. (MIRA 12:9)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut iskusstvennogo
volokna.

(Viscose)

MIKHAYLOV, M.M.; PLATONOVA, V.A.

Changes in the bronchial tree in chronic nonspecific pneumonia in children; clinicobronchographic comparisons. Sov. med. 26 no.4:91-94 Ap '63. (MIRA 17:2)

1. Iz kafedry gosptal'noy pediatrii (zav. -- kand. med. nauk V.P. Sitnikova) i kafedry rentgenologii s meditsinskey radiologiyey (zav. -- dotsent M.M. Mikhaylov) Voronezhskogo meditsinskogo instituta.

PLATONOVA, V., mladshiy nauchnyy sotrudnik

They added 11,000 rubles to the profit of their collective farm.
Zashch. rast. ot vred. i bol. 10 no. 9:4-6 '65.

(MIRA 18:11)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut zashchity
rasteny.

KUZOVKOV, A.D.; PLATONOVA, T.F.

Aconitic alkaloids. Part. 18: Structure of isothalatisdine, talatisidine, condelfine, and monoacetylsongorine. Zhur. ob. khim. 31 no.4:1389-1399 Ap '61. (MIRA 14:4)

1. Vesosyusnyy nauchno-issledovatel'skiy khimiko-farmatsevticheskiy institut imeni S. Ordzhonikidze.
(Alkaloids)

KUZOVKOV, A.D.; PLATONOVA, T.F.

Aconitic alkaloids. Part 19: Structure of eldeline and delpheline.
Zhur.ob.khim. 32 no.4:1290-1293 Ap '62. (MIRA 15:4)

1. Vsesoyuznyy institut lekarstvennykh i aromaticeskikh rasteniy.
(Eledeline) (Delpheline)

PLATONOVA, T.F.; MASSAGETOV, P.S.; KUZOVKOV, A.D.

Lallemantin a new alkaloid from the plant Lallemantia peltata
(L.) Fisch. et Mey. Med. prom. 16 no.2:14 F '62. (MIRA 15:3)

1. Vsesoyuznyy nauchno-issledovatel'skiy khimiko-farmatsev-
ticheskiy institut imeni S. Ordzhonikidze.

(LALLEMANTIA)

(ALKALOIDS)

PLATONOVA, T.F.; KUZOVKOV, A.D.

Alkaloids from the seeds of the plant *Delphinium orientale* J. Gay.
Med.prom.17.no.4:19-20 Ap '63. (MIRA 16:7)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut lekarstvennykh
i aromaticeskikh rasteniy.
(ALKALOIDS) (LARKSPUR)

BA

BA
22

Adsorption method of isolating alkaloids. A. A. Kononova, T. F. Flatonova, and R. A. Kononova (*J. appl. Chem., USSR*, 1960, 33, 876-881).—Bentonite activated by warming with H_2SO_4 was used for adsorption of scopolamine (I) and scopolidone (II) from their solutions and from natural juices containing these alkaloids obtained by treatment of *Salvia ribkovi* with 1% H_2SO_4 . 20% bentonite

adsorbs completely both alkaloids from their 1% solutions in strongly or weakly acidic or weakly alkaline solutions (HCl and H_2SO_4 were used as acids). Experiments were carried out at pH 0.86, 6.1, and 8.8. Adsorption from natural juices is much lower. It is necessary to use 25-30% of bentonite for complete adsorption of I and II from their 0.3% solution, whilst for adsorption from 0.6% solution of pure I only 10% of bentonite is necessary. $CHCl_3$ with NH_3 desorb 71.4-81.6% of I (desorption increases with increased content of NH_3 in $CHCl_3$, up to 7% of NH_3) and 85-87% of II. $3N-NaOH$ and $N-H_2SO_4$ are not to be recommended for elutriation. Elutriation with $CHCl_3$ and NH_3 is advantageous also because I and II may be afterwards easily separated due to the high solubility of I compared with II in $CHCl_3$.

J. B. J. ZABA.

NAME DOV, G.H.; FLATONOVA, T.F.; KUCYKOV, A.D.

*Preparation of a new, symmetrical (C₁₂H₁₀O₂)₂ derivative, growing
in the Doka Museum in 1951, and in the USSR. (See also 1951-1952)*
(1951-1952)

1. Vsesoyuznyy Institut Toksikologiykh i Aromaticheskikh Reagentov
i Institut Botaniki AN AzerSSR

PLATONOVA, S.G.

Heat transfer and resistance of some compact surfaces. Izv.-fiz.
zhur. 7 no.9:3-9 S '64. (MIRA 17:12)

1. Politeknicheskii institut im. M.I.Kalinina, Leningrad.

L 5304-66 EWT(1)/EWA(j)/EWT(m)/EWA(b)-2 JK/RM

ACC NR: AP5025007

SOURCE CODE: UR/0286/65/000/016/0067/0067

AUTHORS: ^{44,55}Platonova, T. F.; ^{44,55}Kuzovkov, A. D.; ^{44,55}Khryashcheva, K. M.; ^{44,55}Labzina, L. D.

ORG: none

TITLE: A method for purifying tetracycline. ⁷ Class 30, No. 173888 [announced by the All-Union Scientific Research Institute of Antibiotics (Vsesoyuznyy nauchno-issledovatel'skiy institut antibiotikov)] ^{44,55}

SOURCE: Byulleten' izobretaniy i tovarnykh znakov, no. 16, 1965, 67

TOPIC TAGS: ^{44,55}antibiotic, tetracycline, epitetracycline, calcium chloride

ABSTRACT: This Author Certificate presents a method for purifying tetracycline. To remove epitetracycline and other organic admixtures, the solution of the antibiotic is treated with a calcium salt such as calcium chloride at pH of 3.1-3.4, and the resulting compound is washed with water at pH of about 3.0.

SUB CODE: ^{44,55}GC,LS/ SUBM DATE: 28Oct64/ ORIG REF: 000/ OTH REF: 000

OC
Card 1/1

UDC: 615.45.779.931

09010543

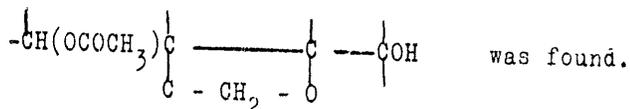
VIZNER, K.; KUZOVKOV, A.D.; PLATONOVA, T.P.

Acetidine alkaloids. Part 20: lebeloline, its structure and transformations. Zhur. ob.khim, 34, no. 5:1666-1668 My '64.
(MIRA 17:7)

1. Vsesoyuznyy institut lekarstvennykh i aromaticeskikh rasteniy (VILAR).

Investigation of the Aconite Alkaloids. XV. On the Structures of Elatine, Methyl Licaconitine, Ajacine, Delsamine, Awadonarine, Licaconitine, and Eldeline SIV/79-29-2-75/8

eldelidine; it is identical with deltaline. In eldeline the grouping



There are 16 references, 8 of which are Soviet.

ASSOCIATION: Vsesoyuznyy nauchno-issledovatel'skiy khimiko-farmatsevticheskiy institut imeni S. Ordzhonikidze (All-Union Scientific Chemopharmaceutic Research Institute imeni S. Ordzhonikidze)

SUBMITTED: July 5, 1958

Card 3/3

Investigation of the Aconite Alkaloids. XV. On the SOV/79-29-8-75/81
Structures of Elatine, Methyl Licaconitine, Ajacine, Delsemine, Awadchari-
dine, Licaconitine, and Eldeline

the position of the esterified hydroxyl was determined in anthranoyllicoctonine as well as its derivatives (methyllicaconitine, delsemine, ajacine, awadcharidine, and licaconitine). Thus in all the above alkaloids the primary hydroxyl group of licoctonine is esterified by anthranilic acid. These alkaloids differ amongst each other by the nature of the radical which arylates the amino group of anthranilic acid. The alkaloid ajacine corresponds, as is known, to the compound (VI) (Ref 11). The structural investigation of methyllicaconitine corresponds to the one made by R. C. Cookson and co-workers. It is thus represented by formula (VII). The alkaloid delsemine (Refs 8,12) corresponds to formula (IX). The alkaloid awadcharidine corresponds to structure (X), since (according to studies by S. Yu. Yunusov and N. K. Abubakirov (Ref 8)) it differs from delsemine by the fact only that it is a derivative of succinic acid while delsemine is a derivative of methylsuccinic acid. For licaconitine formula (XII) was found. The alkaloid eldeline is an acetic ester of the amino alcohol

Card 2/3

5(3)

SOV.79-29-8-73/81

AUTHORS: Kuzovkov, A. D., Platonova, T. F.

TITLE: Investigation of the Aconite Alkaloids. XV. On the Structures of Elatine, Methyl Licaconitine, Ajacine, Delsemine, Awadcharidine, Licaconitine, and Eldeline

PERIODICAL: Zhurnal obshchey khimii, 1959, Vol 29, Nr 8, pp 2782 - 2786 (USSR)

ABSTRACT: Of the otherwise well investigated polyhydroxylated alkaloids of the plant genera Aconitum and Delphinium the structures of methyl licaconitine, ajacine, delsemine, awadcharidine, and licaconitine which are esters of licoctonine (I) had until recently remained unknown, since it had not been known which of the three hydroxyl groups in them is subjected to an esterification when the ester is formed. In the authors' opinion the question can be solved on the basis of the previously obtained data and the material collected by them experimentally, although on a small scale. For the alkaloid elatine (Ref 4), as an ester of the acid (II), the authors suggest the formula (IV), for its amino alcohol, elatidine, the formula (III).
By transformation of the elatidine into anthrancyllicoctonine

Card 1/3

KUZOVKOV, A.D.; PLATONOVA, T.F.

Possible structure of the alkaloid kondelfin. *Med.prom.* 13 no.9:
12-13 S '59. (MIRA 13:1)

1. Vsesoyuznyy nauchno-issledovatel'skiy khimiko-farmatsevticheskiy
institut imeni S. Ordzhonikidze.
(ALKALOIDS)

SOV/79-28-11-52/55

On the Alkaloids of the Plants of the Family Asclepiadaceae (Piknoid),
I. Antitoxicum Funebre (Boiss. et Ky.) Pobed.

the alkaloids of the latter. Antitoxicum funebre (Boiss. et Ky.) is a plant of a height of 40-70 cm. It grows on rocky slopes and unfertile places. The plants collected in the Caucasus in dried state contained 0.26 % alkaloids. In the paper chromatography the bases R_f 0.45 and 0.33 were found among others, in the aluminum oxide chromatography the base $C_{23}H_{25}O_3N$ (R_f 0.45), which has three OCH_3 groups. In the infrared spectrum the bands of the lactam carboniles and of the groups OH and NH are not present. The alkaloid hitherto not described was given the name antofine. Besides, two bases were separated in very small yields with only their melting points being determined. There are 5 references, 4 of which are Soviet.

ASSOCIATION: Vsesoyuznyy nauchno-issledovatel'skiy khimiko-farmatsevticheskiy institut imeni S. Ordzhonikidze (All-Union Scientific Chemo-Pharmaceutical Research Institute imeni S. Ordzhonikidze)

Card 2/3

SCY/79-28-11-32/53

AUTHORS: Platonova, T. E., Kuzovkov, A. D., Masargotov, P. S.

TITLE: On the Alkaloids of the Plants of the Family Asclepiadaceae (Milkweed) (Ob alkaloidakh rasteniy sem. Asclepiadaceae (lastovnevykh)) I. Antitoxicum Funebre (Boiss. et Ky.) Pobed. (I. Antitoxicum funebre (Boiss. et. Ky.) Pobed.)

PERIODICAL: Zhurnal obshchey khimii, 1958, Vol 28, Nr 11, pp 3131-3133 (USSR)

ABSTRACT: The content of alkaloids of the plant family Asclepiadaceae is low. The authors report on the separation of two alkaloids (Ref 1), the tylophorine $C_{24}H_{27}O_4N$ and the tylophorinine $C_{23}H_{27}O_4N$ from the plant Tilophora asthmatica Wigt a. Arn., as well as of the nicotine from the Asclepias syriaca (Ref 2). Of the 40 types of this plant family growing in the USSR only 2, the Cynanchum acutum and the L. Vincetoxicum sibiricum have been investigated until now; no alkaloids were found in them (Ref 4). The great interest for alkaloids of the plant family Apocynaceae and the botanical similarity with that of the Asclepiadaceae caused the authors to investigate

Card 1/3

SOV/79-24-11-51/55

Alkaloids of Plants of the Family Chenopodiaceae (Goosefoot). Analysis
Jaxartica and Arthropytum Leptocladum

identical with that of indole. Based on the above said as well as on the similarity of the constants of the obtained base and its salts with those of 5-methyl-1,2,3,4-tetrahydro- β -carbinol (Ref 4, Table 2) they can be regarded as identical. This carbinol had never before been found in plants. There are 2 tables and 5 references, 2 of which are Soviet.

ASSOCIATION: Vsesoyuznyy nauchno-issledovatel'skiy khimiko-farmatsevticheskiy institut imeni S. Ordzhonikidze
(All-Union Scientific Chemo-pharmaceutical Research Institute imeni S. Ordzhonikidze)

SUBMITTED: October 15, 1957

Card 3/3

SCN/79-94-11-11/35

Alkaloids of Plants of the Family Chenopodiaceae (Goosefoot). Analysis
Jaxartica and Arthrophytum leptocladum

Following absorption bands can be seen: 3.0 μ (a weak band, (NH)), 3.8 μ (a double band, hybrid ion), 5.2 μ (a double band, conjugated bonds). Their nitrogen is of secondary type. The composition and the properties of base Nr 1 as well as the constants of its salts correspond rather exactly to those of N-methyl-4-oxy- β -phenyl-ethyl amine (Table 1) which never before had been found on plants, although its methyl derivative (hordenine - Ref 2) has been long known. The base Nr 2 $C_{12}H_{15}ON$ differs from base Nr 1 by the group CH_2 . Its spectrum differs very little from that of base Nr 1. The authors called this new base Jaxartinin. From Arthrophytum leptocladum M. Pop. N. K. Yurashevskiy (Ref 3) obtained dipterine, leptocladine, and N-methyl- β -phenyl-ethyl amine. It contained 3.7% bases. The authors succeeded in additionally separating the base $C_{12}H_{14}N_2$ which has an NCH_3 group. One of the nitrogen atoms is of basic character. This base with an excess hydrochloric acid forms a chlorine monohydrate. Its infrared spectrum points to an associated NH-group. The ultraviolet spectrum is

Card 2/3

007/79-28-11-31 51

AUTORS: Platonova, T. F., Kuzovkov, A. D., Masburev, I. S.

TITLE: Alkaloids of Plants of the Family Chenopodiaceae (Goosefoot)
(Alkaloidy rasteniy semeystva Chenopodiaceae (narevyye))
Anabasis Jaxartica and Arthrophytum Leptocladum (Anabasis
jaxartica i Arthrophytum leptocladum)

PERIODICAL: Zhurnal obshchey khimii, 1958, Vol 28, Nr 11, pp 3128-3131
(USSR)

ABSTRACT: Anabasis jaxartica (Bge.) Benth. is a 20-30 cm high, perennial
plant, which grows in the salt plains of the Syr-Dar'ya river.
The dry plant contains 0.16 % alkaloids. In alkaloid mixtures
four bases were produced by paper chromatography (R_f 0.30,
0.41, 0.17). In the separation, depending on the basicity,
it was possible to separate 2 bases, Nr 1 (R_f 0.33) and Nr 2
(R_f 0.41). The substances with the denominations R_f 0.30 and
0.17 on the chromatogram yield weak stains and obviously are
contained in the plant in only small quantities. Base Nr 1
corresponds to the formula $C_9H_{13}ON$; it is water soluble and
is dyed with iron chloride. In the infrared spectrum the fol-

Card 1/3

SOV/79-28-11-50/58

Investigation of the Aconite Alkaloids. XIII. Alkaloid *Aconitum Reichenii-*
folium Kar. et Kir. and *Aconitum Nemorum* M. Pop.

tubers grown together. For the investigation the whole plants were collected in August during the blossom in the Tyan'-Shan, in the area of Lake Son-Kul'. The dried plant contained 0.2 % alkaloids. Two bases found in the mixture by paper chromatography they turned out to be talat isamine (Ref 2) and monoacetyl talat isamine. The latter was also obtained by the acetylation of the former (Ref 3); it had never before been found in plants. There are 3 Soviet references.

ASSOCIATION: Vsesoyuznyy nauchno-issledovatel'skiy khimiko-farmatsevticheskiy institut imeni S. Ordzhonikidze (All-Union Scientific Chemo-Pharmaceutical Research Institute imeni S. Ordzhonikidze)

SUBMITTED: October 15, 1957

Card 2/2

SOI/19-05-11-30/55

AUTHORS: Platonova, T. F., Kuzovkov, A. D., Massagatov, I. S.

TITLE: Investigation of the Aconite Alkaloids (Issledovaniye akoni-
lovykh alkaloidov) XIII. Alkaloid Aconitum Rotundifolium Kar.
et Kir. and Aconitum Nemorum M. Pop. (XIII. Alkaloidy Aconitum
rotundifolium Kar. et Kir. i Aconitum nemorum M. Pop.)

PERIODICAL: Zhurnal obshchey khimii, 1958, Vol 28, Nr 11, pp 3126-3128
(USSR)

ABSTRACT: The orbicular-leaf aconite (Aconitum rotundifolium Kar. et
Kir.) is a small perennial plant with two tubers and pale-
lilac flowers. The underground parts of it were collected
during blossom in August and September in the central part of
Tyan'-Shan for the purpose of separating the alkaloids. The
dried plant consisted of about 0.15 % alkaloids. In the mix-
ture four bases were found by paper chromatography. In the
aluminum oxide chromatography alkaloids of the empiric formu-
lae $C_{27}H_{31}O_6N$ and $C_{26}H_{34}O_2N_2$ were separated from the mixture.
The forest aconite (Aconitum nemorum M. Pop.) is a perennial
grass-type plant with blue flowers and a little chain of

Card 1/2

TERLO, G.Ya.; IZRAL'YANTS, Ye.D.; MANTO, Ye.B.; PLATOVA, T.F.

Selecting efficient formulas for antifouling paints with long
action. Lakokras. mat. i ikh prim. no.5:6-10 '63. (MIRA 16:11)

70-1-56/63

Investigation of Aconite Alkaloids. IX. On the Alkaloids of *Aconitum excelsum*

dry roots contain 0,016 % aksin and 0,002 % aksinatin. In this manner the investigated material contained at least five bases. The products R_b 0,49 and 0,38 could not be separated as such. Aksin, C₂₁H₂₉⁸O₅N, contains two alcohol groups and one acetoxy group and apparently also a single ether bond. Aksinatin, C₂₁H₂₇⁴O₄N, has an alcohol and a keto group and, like aksin, also an acetoxy group. By saponification of the latter amino alcohols were obtained from both, accordingly - aksinidine, C₁₉H₂₇⁴O₄N, and aksinatidine, C₁₉H₂₅²⁵O₃N. The superterrestrial parts of the plant which were collected in the pre-flowering-time contained 0,5 % of the noncrystalline alkaloid sum; by paper chromatography of the sum three products were determined (R_b 0,64, 0,49, 0,38), from which only lappaconitine (0,06 %) could be liberated. There are 2 references, all of which are Slavic.

ASSOCIATION: All-Union Scientific Chemical-Pharmaceutical Institute imeni S. Ordzhonikidze (Vsesoyuznyy nauchno-issledovatel'skiy khimiko-farmatsevticheskiy institut imeni S. Ordzhonikidze)

SUBMITTED: January 6, 1957

AVAILABLE: Library of Congress

Card 2/2

1. Chemistry 2. Flora-Chemical analysis

PLATONOVA, T. F.

75-1-56/43

AUTHORS: Platonova, T. F. , Kuzovkov, A. D. , Massagetov, P. S.

TITLE: Investigation of Aconite Alkaloids (Issledovaniye akonitovykh alkaloidov) IX. On the Alkaloids of Aconitum excelsum (IX. Ob alkaloidakh Aconitum excelsum)

PERIODICAL: Zhurnal Obshchey Khimii, 1958, Vol.28, Nr 1, pp.258-261 (USSR)

ABSTRACT: The alkaloids of the plant Aconitum excelsum Rehb. were investigated by S. Y. Yunusov in whose report a short indication to the isolation of mesaconitine and two bases. These bases were characterized by the melting points (265 - 267° and about 100°C). The authors investigated the roots of the plant. The material was collected in Tyan'Shane in fall after the dying of their upper parts. The present sample contained about 3% alkaloid mixture whose chromatographic investigation on paper only indicated three products (R_f 0,66 , 0,49 and 0,38). By splitting up the mixture it was possible to liberate lapaconitine (R_f 0,66) and two new alkaloids (R_f 0,74 and 0,76) which it was not possible to determine chromatographically and for whom the names aksin and aksinatin were suggested. The

Card 1/2

PLATONOVA, T. F.

Alkaloids of plants of the *Ruscaceae* family. Isolation of tetrahydroarmol and *N*-methyltetrahydroarmol from the bark of *Ruscus angustifolius*. T. F. Platonova, A. D. Kuzovkov, and P. S. Maslennikov (E. Orosankin Institute of Chemistry, Pharm. Research Inst., Moscow). *Zh. Obshch. Khim.* 26, 3250-3 (1956); *C.A.* 30, 6724; Maslennikov, *ibid.*, *C.A.* 31, 2490. — Expt. 83 kg. of the bark with 60% EtOH contg. 1% AcOH, dig. with 2 vols. of H₂O and evap. to 20 l. gave an alkaloid-free fatty ppt. which was reprec. The filtrate treated with 25% NH₄OH gave a flocculent ppt. from which no individual substance was isolated. The soln. acid. with HCl, the ext. treated with aq. H₂SO₄ and the solids ext. treated with NH₄OH and acid. with HCl yielded 11 g. cryst. product, while the filtrate extd. with H₂O gave 112 g. amorphous matter. Chromatography these on paper with BuOH-H₂O-AcOH showed 3 bases with *R*_f 0.36, 0.47, and 0.58. The cryst. product was identified as tetrahydroarmol. *C₁₇H₂₁NO₂*, m. 226°; HCl salt, m. 235°. *R*_f 0.58. The amorphous mat. with 10% H₂SO₄ gave a cryst. sulfate of hexamine-free base, m. 178-80°; *R*_f 0.58, while the residual soln. yielded 6 g. phenolic substances, which treated with 4% HCl gave *N*-methyltetrahydroarmol. *C₁₇H₂₁NO₂*, m. 248-70°; *R*_f 0.47 (HCl salt, m. 274-5°). Harmol with Na in EtOH gave tetrahydroarmol, identical with the above specimen, m. 250°; HCl salt, m. 235°. This with Me₂COH gave the above-described *N*-Me deriv., identified chromatographically, *R*_f 0.47. Infrared and ultraviolet spectra of the alkaloids were given. G. M. Kosoloff.

PLATONOVA, T. MASSA GETOV, P. . . .
fumvilline, $C_{20}H_{19}O_5N$, m. 180.5-1.5° (from Me₂CO),
 $[\alpha]_D -44.4^\circ$ (HCl salt, m. 212°), which contains 2 MeO
 groups and is nonphenolic. *Papaver hybridum* gave 0.12%
 total alkaloids from which was isolated 0.04% *papybrine*
di-HBr salt, decomp. 204° (from H₂O), $[\alpha]_D 44.0^\circ$; *free*
base, $C_{22}H_{20}O_4N_2$, yellow amorphous solid (HI salt, decomp.
 200°). *Roemeria hybrida* gave 0.8% alkaloids, from which
 some protopine was isolated, along with the new *roemeridine*,
 $C_{21}H_{20}O_5N$, m. 228-30°, which contains 11O and 3 MeO
 groups; HCl salt, m. 258-60°. An unknown alkaloid, m.
 230°, was also found. *P. pavoninum* gave 0.1% total
 alkaloids which yielded protopine, α -alloecryptopine (m.
 187-9°), and *roemeridine*, m. 228-8°. *Chelidonium majus*
 gave 0.7% total alkaloids from which were isolated (through
 the sulfates): *sanguinarine*, m. 240-2°; *chelerythrine*, m.
 210-13°; *chelidamine*, $C_{21}H_{19}O_4N$, m. 204-5°, $[\alpha]_D -310.0^\circ$
 (HCl salt, m. 254-6°; *methiodide*, m. 276°). Some pro-
 topine was isolated from the mother liquor after removal
 of chelidamine from MeOH-CHCl₃ soln., and finally
 some *chelidonine*, m. 133°, $[\alpha]_D 110^\circ$, $C_{20}H_{19}O_4N$; HCl salt,
 m. 286°.

C. M. Kosolapoff

2/2

PLATONOVA, T. F.

✓ Alkaloids of plants of the poppy family. T. F. Platonova,
 P. S. Massagetov, A. D. Kuzovkov, and L. M. Utkin
 (S. Ord. Bonikidze All-Union Chem.-Pharm. Sci. Re-
 search Inst., Moscow). *Zhur. Obshchei Khim.* 26, 173-
 80; *J. Gen. Chem. U.S.S.R.* 26, 181-6 (1956) (Engl. trans-
 lation).—Conventional $\text{NH}_4\text{OH}-(\text{CH}_2\text{Cl})_2$ treatment of 14
 kg. dried roots of *Fumaria schleicheri* gave 23 g. Et_2O -sol.
 alkaloids which were solid and 20 g. noncryst. material,
 while 17 g. noncryst. material was extd. with CHCl_3 .
 Chromatography of the 1st group in CHCl_3 over Al_2O_3
 with elution by CHCl_3 -2% MeOH gave 3 g. protopine,
 m. 205-6° (from the colorless zone), 0.2 g. tarry material
 from the orange zone. The yellow zone gave a mixt.
 which on rechromatographing with elution with CHCl_3
 and MeOH gave 3 g. protopine and 2 g. *fumaridine*, yellow
 solid, m. 190-1°, $\text{C}_{12}\text{H}_{16}\text{O}_2\text{N}_2$, which has 2 MeO and 2
 MeN groups; *tartrate*, m. 217°; *di-HCl salt*, m. 213-14°;
picrate, m. 204-5°; *methiodide*, m. 247-8°. The MeOH-
 sol. fractions yielded 1.2 g. *fumaramine*, $\text{C}_{10}\text{H}_{12}\text{O}_4\text{N}_2$, m.
 223-4°, which has 2 N-Me groups; *tartrate*, m. 200°; *HBr*
salt, m. 258-60°. Similar chromatography of the noncryst.
 material gave 0.78 g. *fumarimine*, $\text{C}_{10}\text{H}_{10}\text{O}_4\text{N}$, m. 189-90°
 (contains 1 N-Me group) (*oxalate*, m. 213°; *HCl salt*, m.
 255-7°), some protopine, and 0.3 g. *fumariline*, $\text{C}_{10}\text{H}_{12}\text{O}_4\text{N}$,
 m. 157-9° (contains 1 N-Me group) (*HCl salt*, m. 224°;
HBr salt, m. 219°). Ultraviolet spectra of the new alka-
 loids were examd.; *fumaramine* apparently contains a
 carbonyl group as does *fumaridine*; the latter may be the
 di-Me ether of the former. Similar treatment of roots of
F. micrantha yielded protopine, *fumaramine*, and unidenti-
 fied material. Roots of *F. vaillantii* gave protopine, *fumari-*
dine, and the colorless chromatographic zone yielded

chem 4

1/2

Platonova, T. F.

Alkaloids of plants of *Leontice awaramanii*. II. Prod.

6) **Products of decomposition of thaspine.** T. F. Platonova and A. D. Kuzovkov (S. Ordzhonikidze All-Union Sci. Research Chem. Pharm. Inst., Moscow). *Zhur. Obratsh. Khim.* 24, 2240-50 (1954); cf. *C.A.* 48, 12788f. — Distn. of thaspine with Zn dust in H gave fluorene and a mixt. of amines, from which was isolated a *picrate*, m. 225-6° (from EtOH), corresponding to $C_{17}H_{19}O_2N$. Fusion of thaspine with KDH gave *3,3',x,x'-tetrahydroxybiphenyl*, m. 217.5-18.5°; methylation with CH_3N_3 gave a *tetramethoxybiphenyl*, m. 100.5-1.5°, and some *trimethoxyhydroxybiphenyl*, m. 127.5-8.5°. Acetylation of the tetra-HO compd. with AcCl gave the *tetra-Ac deriv.*, m. 87-8.5°, which oxidized with CrO_3 -AcOH to 2,3-dihydroxybenzoic acid, m. 206-6°. Distn. of the tetra-HO compd. with Zn dust gave *Pb*. Refluxing thaspine with HBr (d. 1.48) 3 hrs. gave *thaspinic acid*: *HBr salt*, decomp. 208°, $C_{17}H_{19}O_2NBr$; *free acid*, amorphous. Oxidation of the free acid with $KMnO_4$ - H_2SO_4 gave β -*dimethylaminopropionic acid* (I), m. 143-4° [*HCl salt*, m. 186-8°; *reineckate*, m. 164-75°; *Et ester*, bp 82-1° (*HCl salt*, m. 143-5°; *meliodide*, m. 178°)]. Thus thaspinic acid, as possibly thaspine, has a $Me_2NCH_2CH_2$ side-chain. Thus the alkaloid is $C_{17}H_{19}(OMe)_2(CH_2O_2)(CO_2)NMe_2$. The oxidation of thaspinic acid, described above, gives directly the Et ester of I, which can be saponified to I with 5% NaOH at 100°.

G. M. Köslovoff

①

Platonova, T. F.

4

~~Alkaloids of plants. Leontice evermannii. I. New
alkaloid saipine and alkaloid isoleontine. Preparation of
leontidine and isoleontine. T. F. Platonova, A. D.
Kunshay, and P. B. Maslennikova. *Trudy Khim. U.S.S.R.*
23, 621-6 (1965) (Engl. translation). See C.A. 46, 8987c.
H. L. H.~~

2/2 Platonova, T. F., et al

382° (from H₂O); methiodide, decomp. 307°; picrate, m.
177-9°. On hydrogenation of IV over Pt O, IV took
up 1.9 moles H₂ and gave an optically active product devoid
of secondary N atoms, C₁₁H₁₅N₃, m. 77-8.5° (from Et₂O);
[α]_D -30.1° (EtOH); picrate, m. 256° (with decompn.);
di-HCl salt monohydrate, m. 368° (with sublimation); meth-
iodide, decomp. 292° (from dil. EtOH).
G. M. K.

PLATONOVA, T. F.

Chemical Abst.
Vol. 48
Apr. 10, 1954
Organic Chemistry

Alkaloids of plants *Leontice evermannii*. I. New alkaloid *leontidine* and alkaloid *isoleontine*. Preparation of *leontidine* and *isoleontine*. T. F. Platonova, A. G. Kuznetsov, and P. S. Maslov. *Dokl. Akad. Nauk SSSR*, 1953, 166, 1063. *Zhur. Obshch. Khim.*, 1953, 29, 230-4 (1953).—Extn. of the upper parts of *L. evermannii* with (CH₂Cl)₂ in the presence of 10% NH₄OH and treatment of the ext. with 15% H₂SO₄ gave a ppt. of *leontine* (I) sulfate, purified by crystn. from 10% AcOH; the yield was 0.18% of the plant wt. The acidic filtrate, made alk. with NH₄OH with cooling and extd. with CHCl₃, yielded 0.66% (on plant wt.) mixed bases as a dark oil. This (830 g.) in 500 ml. Me₂CO acidified to Congo red with alc. HCl gave 270 g. *leontidine-HCl*, m. 203° (from EtOH). The mother liquor after evapn. was distd. *in vacuo*, yielding *pachycarpine*, isolated as the HI salt, m. 233-4°; *leontidine-HI*, m. 189-90° (perchlorate, m. 212-13°; [α]_D -47.8°); *isoleontine*, isolated as the picrate, m. 177-8°; and *leontidine-HCl*, m. 310-11°. I sulfate with 10% NH₄OH and CHCl₃ gave, on addn. of Me₂CO to the org. ext., the free base, C₂₀H₃₀O₂N of I, m. 370°, optically inactive; methiodide monohydrate, needles (from H₂O). Hydrolysis of I with boiling 0.1N NaOH in 15 min. gave an amino acid, C₁₀H₁₆O₂N₂, probably contg. a lactone group, since on heating with dil. H₂SO₄ it forms I. I contains 2 MeO, 1 O-CH₂-O, and a NMe₂ groups. I sulfate, 1.0.5H₂SO₄.H₂O, m. 360° (from H₂O). *Leontidine-HCl*, m. 310-11° (from EtOH), with NaOH gave free *leontidine*, C₁₄H₁₈ON₂, (II), m. 118-19° (from C₆H₅-Et₂O); [α]_D -192.2° (MeOH). On hydrogenation over PtO₂ II takes up 4.2 moles H₂, yielding an oily product, which gave the HI salt, m. 219-20°, [α]_D 10.33°. This new base is C₁₄H₂₄N₂, *leontidine* (III), oil, b_p 141-4°, [α]_D -7.76° (without solvent). II therefore is not a sparteine-group alkaloid and is based on O-free III, which is a homolog of sparteine differing from it by a CH₂ group. This indicates further that *leontamine* is C₁₄H₂₂N₂. III forms a sulfate, m. 229-30°; di-HI salt, decomp. 208° perchlorate, m. 234°, [α]_D 20.55°. *Isoleontine* picrate treated with 18% HCl, washed with dil. HCl and the acid soln. treated with NaOH after removal of picric acid gave free *isoleontine*, C₁₄H₂₀ON₂, (IV), b_p 175-90°, m. 107-8° (from EtO); [α]_D -78.2° (EtOH); perchlorate, decomp.

PLATONOVA, T.A.

Parasites of some fishes of lake Sevan. Paraz. sbor. 21:209-220
'63. (MIRA 17:11)

1. Kafedra zoologii bespozvonochnykh Leningradskogo gosudarstvennogo
universiteta.

PLATONOVA, T.A.

New species of nematodes of the genus *Pseudocella* Filipjev
from the Kurile Islands and southern Sakhalin. Issl.dal'nevost.
mor.SSSR no.8:200-218 '62. (MIRA 15:12)

1. Zoologicheskii institut AN SSSR.
(Kurile Islands--Nematoda) (Sakhalin--Nematoda)

PLATONOVA, T.A., aspirant

Nematodes of the family Leptosomatidae from the region of
Kerguelen Islands. Inform.biul.Sov.antark.eksp. no.3:59-61
'58. (MIRA 12:4)

1. Zoologicheskiy institut AN SSSR.
(Kerguelen Islands--Nematoda)

ИПЛАТОНОВА, С.Т.

21(4) PHASE I BOOK EXPLOITATION SOV/2583

International Conference on the Peaceful Uses of Atomic Energy, 2nd, Geneva, 1958.

Doklady sovetskikh uchenykh; vvedeniye reaktory i vvedeniye ener- getiki. (Reports of Soviet Scientists: Nuclear Reactors and Nuclear Power) Moscow, Atomizdat, 1959. 707 p. (Series: Its: Trudy, vol. 2) Errata slip inserted. 6,000 copies printed.

General Eds.: M. A. Dollezhal, Corresponding Member, USSR Academy of Sciences, A. K. Krasin, Doctor of Physical and Mathematical Sciences, A. I. Lappin, Doctor of Physical and Mathematical Sciences, I. I. Morozov, Corresponding Member, USSR Academy of Sciences, I. V. Fursov, Doctor of Physical and Mathematical Sciences, and V. S. Alyab'yev; Tech. Ed.: Ye. I. Mazel'.

NOTE: This book is intended for scientists and engineers engaged in reactor designing, as well as for professors and students of higher technical schools where reactor design is taught.

COVERAGE: This is the second volume of a six-volume collection on the peaceful use of atomic energy. The six volumes contain the reports presented by Soviet scientists at the Second International Conference on the Peaceful Uses of Atomic Energy, held from September 1 to 13, 1958, in Geneva. Volume 2 consists of three parts. The first in the series is devoted to the construction of reactors in the Soviet Union; the second part is devoted to the construction of reactors in other countries; and the third, which is predominantly theoretical, is devoted to nuclear reactor physics and construction engineering. V. V. Morozkin is the science editor of this volume. See SOV 2021 for titles of all volumes of the set. References appear at the end of the articles.

Mostovoy, V. I., V. S. Dikarev, M. B. Yegizarov, and Yu. S. Saltzyov. Measuring Neutron Spectra in Uranium Water Lattices (Report No. 2152) 546

Krasin, A. K., B. G. Dubovskiy, M. M. Lantsov, Yu. Yu. Glazkov, K. K. Kocharov, A. V. Kamyayev, L. A. Gerasova, V. V. Vavilov, Ye. I. Inyutin, and A. P. Senchenkov. Studying the Physical Characteristics of a Beryllium-moderator Reactor (Report No. 2146) 555

Galanin, A. D., S. A. Nemirovskaya, A. P. Rudik, Yu. G. Abov, V. P. Beikin, and P. A. Krupchitskiy. Critical Experiment on an Experimental Heavy-water Reactor (Report No. 2036) 570

Karchuk, G. I., V. Ya. Puzko, Ye. I. Pogdalina, V. V. Szolov, I. P. Tyaterev, S. T. Platocova, and G. Zhurav. Calculations of Neutron Fluxes in Nuclear Reactor Physics and Methods of Calculating Them (Report No. 2151) 588

Sinyutin, O. V. and V. N. Semenov. Determination of Control Rod Effectiveness in a Cylindrical Reactor (Report No. 2409) 611

Gel'fand, I. M., S. M. Farnberg, A. S. Prolov, and N. M. Cherkasov. Using the Monte Carlo Method of Random Sampling for Solving the Kinetic Equation (Report No. 2141) 628

Lalestin, M. I. Neutron Distribution in a Heterogeneous Medium (Report No. 2189) 638

Kazemovskiy, M. V., A. V. Stepanov, and P. L. Shapiro. Neutron Thermalization and Diffusion in Heavy Media (Report No. 2143) 641

Veynik, A. I., V. S. Yermakov, and A. V. Lykov. Using the Operator Theory for Studying Neutron Diffusion in the Absorbing Media of Nuclear Reactors (Report No. 2224) 668

Broder, D. L., S. A. Kurkin, A. A. Rytuzov, V. V. Levin, and V. V. Orlov. Studying the Spatial and Energy Distribution of Neutrons in Different Media (Report No. 2147) 678

Dmitriyev, A. B. Boron Ionization Chambers for Work in Nuclear Reactors (Report No. 2084) 690

Kirillin, V. A., and S. A. Ulybin. Experimental Determination of Specific Volumes of Heavy Water in a Wide Temperature and Pressure Range (Report No. 2471) 698

PLATONOVA S. P.

MARCHUK, G. I., PUPKO, V. Y., POGUDALINA, E. L., SMELOV, V. V., TUTEREV,
I. P., PLATONOVA, S. P. and DRUZHININA, G. I.

"Nuclear Reactor Physical Problems and Calculations. Methods."

paper to be presented at 2nd UN Intl. Conf. on the peaceful uses of Atomic
Energy, Geneva, 1 - 13 Sep 58.

L 12054-65
ACCESSION NO: APL047438

ENCLOSURE: 02

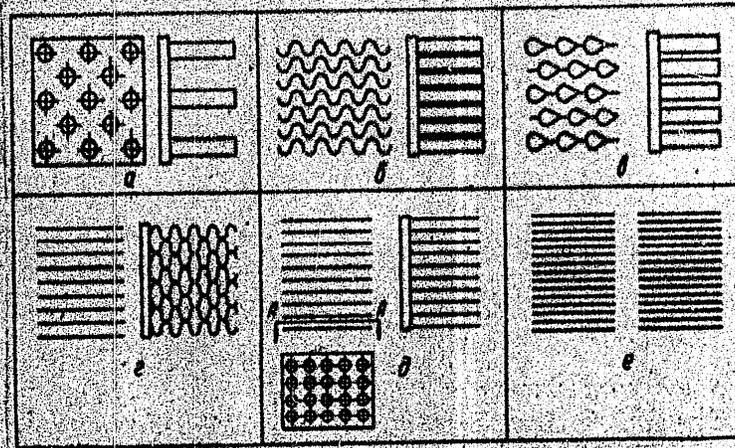


Fig. 2. Types of experimental surfaces.

Card 4/4

L 12054-65
ACCESSION NR: AP4047438

ENCLOSURE: 01

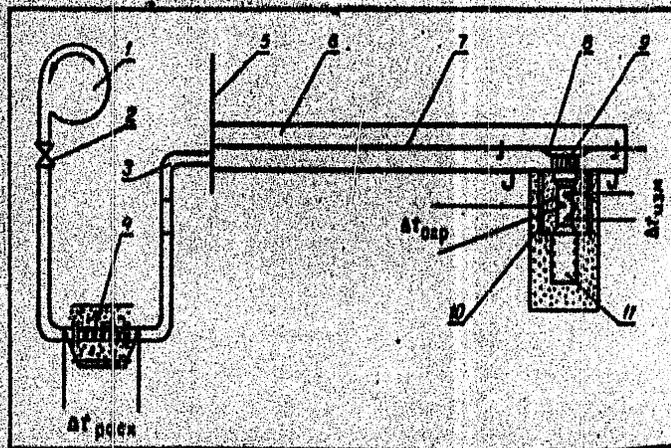


Fig. 1. Schematic of experimental apparatus.

1- fan, 2- baffle, 3- ducting, 4- heater I, 5- vertical baffle plate,
6- rectangular plexiglass duct, 7- horizontal baffle plate, 8- spring,
9- sample, 10- guard, 11- heater II.

Card 3/4

L 12054-65
ACCESSION NR: APL04743B

and

$$\xi = \frac{\Delta P_{\text{loss}}}{\omega_{cr}^2} \frac{2gd_h}{\gamma_s L}$$

were determined for the six surfaces as a function of Reynolds number. It was found that the experimental Nusselt number as a function of Reynolds number could be expressed as $Nu = CR Re^n$ over a range of Re . For the six surfaces $C = 0.207, 0.237, 0.087, 0.043, 0.1, 0.64$ respectively, $n = 0.75, 0.72, 0.73, 0.75, 0.62, 0.41$ respectively, and the equation is valid for $Re = 1000-8000, 1200-6000, 300-3000, 500-2200, 200-1800, 400-3500$ respectively. The flow resistance coefficients ξ varied as follows over the above Re ranges: $0.852-0.669, 0.695-0.44, 0.75-0.49, 0.28-0.15, 0.14-0.035, 0.31-0.065$. An evaluation of the data based on energy loss by the different surfaces showed that pin surfaces with cylindrical pins in a staggered order are most advantageous, but their weight and size characteristics are inferior. Orig. art. has: 3 figures, 3 tables, and 13 formulas.

ASSOCIATION: Politekhnikheskiy institut in. M. I. Kalitina, g. Leningrad (Leningrad Polytechnic Institute)

SUBMITTED: 01Feb64

SUB CODE: TD
Card 2/4

NO REF SOV: 002

ENCL: 02
OTHER: 003

L 12054-65 ENT(1)/ENT(m)/EPF(c)/EPF(n)-2/EPR/T/EMP(t)/EPA(bb)-2/EWP(b)/EWA(1)
 Pr-4/Ps-4/Pu-4 LJP(c)/ASD(f)-2/ASD(s)-5/AFWL/BSO/AEDC(a)/SSD/AS(mp)-2/ESD(t)
 JD/MS
 ACCESSION NR: APL047438
 8/0170/64/000/009/0003/0009

AUTHOR: Platonova, S. G.

TITLE: Investigation of heat transfer and flow resistance of several compact surfaces ² B

SOURCE: Inshenerno-fizicheskiy zhurnal, no. 9, 1964, 3-9

TOPIC TAGS: heat transfer, semiconductor, compact surface, surface heat transfer, flow resistance

ABSTRACT: The experimental apparatus shown in Fig. 1 on the Enclosures was used to measure the heat transfer and flow resistance characteristics of six types of compact surfaces made of red copper, (see Fig. 2 on the Enclosures). These surfaces are of interest for semiconductor devices. The flow Reynolds number could be changed by the fan (1) and heater (4) and the heat transfer to the air stream could be varied by the electric heater (11). The heat flux through the heater (11) and the average logarithmic temperature difference were measured and the heat transfer coefficient determined from these quantities. The dimensionless quantities

$$Nu = \frac{\alpha d_s}{\lambda}$$

Card 1/4

L 12054-65 EWT(1)/EWT(m)/EPF(c)/EPF(n)-2/EPR/T/ENP(t)/EPA(bb)-2/ENP(b)/EWA(1)
 Pt-4/Ps-4/Pu-4 IJP(c)/ASD(f)-2/ASD(a)-5/AFWL/BSO/AEDC(a)/SSD/AS(mp)-2/ESD(t)
 JD/WB
 ACCESSION NR: APL047436 8/0170/64/000/009/0003/0009

AUTHOR: Platonova, S. G.

TITLE: Investigation of heat transfer and flow resistance of several compact surfaces B

SOURCE: Inzhenerno-fizicheskiy zhurnal, no. 9, 1964, 3-9

TOPIC TAGS: heat transfer, semiconductor, compact surface, surface heat transfer, flow resistance

ABSTRACT: The experimental apparatus shown in Fig. 1 on the Enclosures was used to measure the heat transfer and flow resistance characteristics of six types of compact surfaces made of red copper (see Fig. 2 on the Enclosures). These surfaces are of interest for semiconductor devices. The flow Reynolds number could be changed by the fan (1) and heater (4) and the heat transfer to the air stream could be varied by the electric heater (11). The heat flux through the heater (11) and the average logarithmic temperature difference were measured and the heat transfer coefficient determined from these quantities. The dimensionless quantities

$$Nu = \frac{q d_h}{\lambda}$$

Card 1/4

L 38573-65
ACCESSION NR: AR5006200

0

A generalized heat-transfer relation is obtained for several corrugated surfaces, in the form $Nu = f(Re)$. The experimental set-up and the measurement procedure are described. Three illustrations. Bibliography, 3 titles.

SUB CODE: TD

ENCL: 00

02
Card 3/3

L 38573-65

ACCESSION NR: AR5006200

	First surface	Second surface
Spacing of ribs, m.	2.5×10^{-3}	1.5×10^{-3}
Height of ribs, m.	3.2×10^{-2}	4.2×10^{-2}
Total heat transfer surface, m^2 .	710×10^{-4}	1135×10^{-4}
Hydraulic diameter, m.	3.8×10^{-3}	2.7×10^{-3}
Thickness of ribs, m.	0.5×10^{-3}	0.5×10^{-3}
Ratio of heat-transfer area to rib volume, m^2/m^3 .	1390	1690
Ratio of silvered surface to total surface.	0.975	0.982
Diameter of corrugations, m.	2×10^{-3}	1×10^{-3}

The experiments were made with the base of the ribs reaching a temperature up to 100C, and in the range $100 < Re < 3000$. The results obtained in the investigation of the heat transfer are plotted in the form $\log N = f_1(\log Re)$, and the defining dimension is chosen to be the hydraulic diameter of the constricted section of the stream. Data on the resistance are represented in the form $\zeta = f_2(Re)$. A comparison is made of the experimental data and the results of the investigation of corrugated surfaces with different geometrical parameters, carried out by other workers.

Card 2/3

L 38573-65 EWT(1)/IPT(o)/EPT(n)-2/EWG(m)/EPR Pr-4/Pg-4/Pu 4 WW
ACCESSION NR: AR5006200 S/0272/64/000/012/0120/0120

SOURCE: Ref. zh. Metrologiya i izmeritel'naya tekhnika. Otdel'nyy vypusk, Abs.
12.32.760

AUTHORS: Platonova, S. G.

TITLE: Investigation of heat exchange and resistance of some compact heating sur-
faces

CITED SOURCE: Uch. zap. aspirantov i soiskateley. Leningr. politekhn. in-t.
Energomashinostroyeniye. L., 1964, 25-31

TOPIC TAGS: heat transfer, heat exchange, thermal resistance

TRANSLATION: Results are presented of an experimental investigation of two corru-
gated heating surfaces placed in an air stream. The experimental surfaces were
made in the form of wavy ribs with half-round corrugations, soldered with tin to a
red-copper bearing plate. The main geometrical data on the surfaces are presented.

Card 1/3

ILLEGIBLE

SAKHAROV, V.V.; MANSUROVA, V.V.; PLATONOVA, R.N.; SHCHERBAKOV, V.K.

Detection of physiological resistance to ionizing radiation in
autotetraploid plants of common buckwheat field. Biofizika 5
no. 5:558-565 '60. (MIRA 13:10)

1. Institut biologicheskoy fiziki AN SSSR, Moskva.
(PLANTS, EFFECT OF RADIATION ON) (POLYPLOIDY)

Cytological proofs of ...

S/747/62/000/000/024/025
D243/D307

forms and that this situation was unchanged after irradiation. Tetraploids showed a smaller percentage of aberrants after 0.5, 1.0, 5.0 and 10.0 kr of γ radiation. This is discussed in relation to physiological protection and was confirmed by the authors' experiments reported in greater detail elsewhere (Biofizika, 1960, 5, no. 5, 558-569). The 4x forms were shown to be twice as stable as the 2x forms to high speed neutrons. The effect of γ radiation and subsequent storage was examined by storing the seeds for periods of 6 and 12 months after irradiation. After 6 months, the irradiated seeds of both forms showed a regular, steep rise in the percentage of chromosome aberration. This was more marked in the 4x forms. Both forms possess mechanisms which interfere with the conversion of potential into actual chromosome aberrations and these protective mechanisms are particularly effective in tetraploid forms. There are 3 figures and 2 tables.

ASSOCIATION: Institut biologicheskoy fiziki AN SSSR, Moskva (Institute of Biological Physics, AS USSR, Moscow)

Card 2/2

S/747/62/000/000/024/025
D243/D307

AUTHORS: Sakharov, V. V., Mansurova, V. V., Platonova, R. N. and
Shcherbakov, V. K.

TITLE: Cytological proofs of the physiological protection of
autotetraploids of buckwheat (*Fagopyrum esculentum* moench)
from the effect of ionizing radiation

SOURCE: Radiatsionnaya genetika; sbornik rabot. Otd. biol. nauk
AN SSSR. Moscow, Izd-vo AN SSSR, 1962, 346-357

TEXT: The results are summarized of a comparative, cytogenetic
study of the effect of different types of radiation on diploid and
autotetraploid plants of common buckwheat (*Fagopyrum esculentum*
moench), using dormant seeds kept under identical room conditions
for the same period. The higher sensitivity of diploid forms to both
γ and x radiation was confirmed, diploids showing depression of
growth after 10 kr of γ radiation, and the autotetraploids after
30 kr. Cytological examination showed that the percentage of ab-
errant cells in nonirradiated controls was equal (2.2%) in both

Card 1/2

PLATONOVA, R. N., and SAKHAROV, V. V.,

"Selection for Radioresistance and Resistance to Chemical Mutagens in Diploid and Teraploid Forms of Buckwheat (*Fagopyrum esculentum*)."

report submitted for the 11th Intl. Congress of Genetics, The Hague, Netherlands,
2-10 Sep 63

LEVIN, Ya.A.; PLATONOVA, R.N.; KUKHTIN, V.A.

Condensed heterocycles. Report No.8: Condensation of 3-amino-
1,2,4-triazole with cyanoacetic ester. Izv. AN SSSR, Ser. khim.
no.8:1475-1480 Ag '64. (MIRA 17:9)

1. Institut organicheskoy khimii AN SSSR, Kazan'.

FLATONOVA, R.H., SAKHAROV, V.V.

Chemical mutagens and ploidy in plants. *Genetika* 10, 1965
S 165. (USSR)

1. Institut biologicheskoy fiziki AN SSSR, Moskva. Submitted
March 30, 1965.

On the Reaction of Phosgene and Oxalyl Chloride
With Esters of Phosphorous Acid

SOV/79-29-2-31/71

The reaction of the chloric anhydride of oxalic acid with trimethyl and triethyl phosphite also takes place in the same way, yet the respective yields are smaller (last-mentioned Scheme). There are 2 tables and 9 references, 7 of which are Soviet.

ASSOCIATION: Kazanskiy gosudarstvennyy universitet (Kazan' State University)

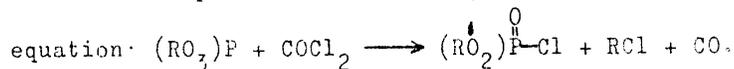
SUBMITTED: December 26, 1957

Card 3/3

On the Reaction of Phosgene and Oxalyl Chloride
With Esters of Phosphorous Acid

SOV/79-29-2-31/71

$((C_2H_5O)_2\overset{O}{P}-COOC_2H_5)$ ought to be formed. This ester obtained by A. Ye. Arbuzov (Ref 7) by action (VI) of chlorocarbonic ester on triethyl phosphite was not identical with compound (VI). For this reason, the authors repeated its synthesis according to reference 7. The constants of the products synthesized are listed in table 1. It results from them that compound (V) does not agree with (VI) but represents a triethyl phosphate. After further experiments of identification, i.e., due to the reaction of phosgene with the methyl, ethyl, and butyl ester of phosphorous acid, and 2) after the synthesis of some chloric anhydrides of dialkyl phosphoric acids by the action of chlorine on dialkyl phosphorous acids (Table 2), in the course of which the reaction products were identical in both cases and represented chloric anhydrides of dialkyl phosphoric acid, the following results were obtained: the reactions of phosgene with complete esters of phosphorous acid take place in any case under separation of the CO group according to this



Card 2/3

AUTHORS: Pudovik, A. N., Platonova, R. K. SOV, 59-59-2-31/73

TITLE: On the Reaction of Phosgene and Oxalyl Chloride With Esters of Phosphorous Acid (O reaktsii fosgena i khloristogo oksalila s efirami fosforistoy kisloty)

PERIODICAL: Zhurnal obshchey khimii, 1959, Vol 29, Nr 2, pp 507-510 (USSR)

ABSTRACT: In connection with previous investigations concerning reactions of phosphites with halogen anhydrides of α -halogen-substituted carboxylic acids it was a matter of course that the authors would carry out the reactions of phosphites with halogen anhydrides of dibasic acids. The results they obtained in the reactions of complete phosphites with phosgene were in contrast to those described by M. I. Kabachnik and P. A. Rossiyskaya (Ref 6). On the reaction of phosgene with triethyl phosphite the authors obtained the chloric anhydride (IV), which, together with ethyl alcohol in the presence of pyridine or sodium ethylate, permitted substitution of the ethoxy group for chlorine, while ester (V) was formed simultaneously. According to the data obtained by Kabachnik's and Rossiyskaya's investigation, the ethyl ester of diethyl phosphonoformic acid

Card 1/3

PLATONOVA, R. N.; SAEHAROV, V. V.

Radioreistant forms of dicloni and acetotraploid buckwheat
and its hybrids with common (2x and 4x) buckwheat from its
populations. Radiobiologiya 4 no.4:613-618 '64.

(MIRA 10:11)

1. Institut biologicheskoy fiziki AN SSSR, Moskva.

PLATONOVA, P.I.; PROKOF'YEV, V.K., prof., otvetstvenny red.; TYUMENEVA, S.T.,
red.; FREGER, D.P., tekhn.red.

[Solution technique for the spectrum analysis of sodium and
calcium admixtures in magnesium] Spektral'noe opredelenie primesei
natriia i kal'tsiia v magnii metodom rastvorov. Leningrad, 1955.
7 p. (Leningradskii dom nauchno-tekhnicheskoi propagandy.
Informatsionno-tekhnicheskii listok, no.100(788)) (MIRA 10:12)
(Magnesium--Spectra)

BUKAVTSOVA, V.F.; STIFATOVA, N.N.; KONOBKIN, V.B.; MOROZOVA, T.I.;
SOPRONOVA, V.A.; SHAPOROST, P.D.; PLATONOVA, N.P.; YEREMENKO, G.S.;
IVANOVA, A.M.; SILAYEVA, N.Ya.; SUYETINA, S.M.; PALUYANOV, I.Ye.;

Study of the dust factor in the founding departments of six
Krasnodar plants. Nauch. trudy Kub. gos. med. inst. 19:63-76
'62.
(MIRA 17:8)

1. Iz sanitarno-epidemiologicheskoy stantsii g. Krasnodara
i polikliniki No.8 Krasnodara.

ZAKHAROV, A.A.; ZVER'KOV, B.V.; PLATONOVA, N.G.

Device for testing specimens for long-period strength in bending
in tensile-testing machines. Zav.lab. 28 no.8:1005-1006 '62.
(MIRA 15:11)

1. TSentral'nyy kotloturbinnyy institut imeni I.I.Polzunova.
(Testing machines)

ABRAMOVA, N.A., nauchn. sotr.; VOYEVODSKIY, A.S., nauchn. sotr.;
GINZBURG, O.F., doktor khim. nauk; YERSHOVA, Ye.TS., kand.
khim. nauk; KOLYCHEV, V.B., nauchn. sotr.; MAR'YANOVSKAYA,
K.Yu., nauchn. sotr.; MAZEL', R.L., nauchn. sotr.;
MEL'NIKOVA, N.S., nauchn. sotr.; PLATUNOVA, N.B., nauchn.
sotr.; REMOZOV, A.L., kand. khim. nauk; UTOCHKIN, V.V.,
nauchn. sotr.; KHAVIN, Z.Ya., kand. khim. nauk; EFROS, L.S.,
doktor khim. nauk; NIKOL'SKIY, B.P., glav. red.; RABINOVICH,
V.A., kand. khim. nauk, zam. glav. red.; GRIGOROV, O.N.,
doktor khim. nauk, red.; POZIN, M.Ye., doktor tekhn. nauk,
red.; PORAY-KOSHITS, B.A., doktor khim. nauk, red.;
RACHINSKIY, F.Yu., kand.khim. nauk, red.; ROIANKOV, F.G.,
doktor tekhn. nauk, red.; FRIDRIKHSBERG, D.A., kand. khim.
nauk, red.; ZONIS, S.A., red.; LEVIN, S.S., tekhn. red.;
ERLIKH, Ye.Ya., tekhn. red.

[Handbook of chemistry] Spravochnik khimika. 2. izd., perer.
i dop. Leningrad, Goskhimizdat. Vol.2. [Basic properties of
inorganic and organic compounds] Osnovnye svoistva neorgani-
cheskikh i organicheskikh soedinenii. 1963. 1167 p.

(MIRA 17:3)

1. Chlen-korrespondent AN SSSR (for Nikol'skiy).

KHAZANOVA, G. (Chelyabinskaya oblast'); PLATONOVA, N. (Chelyabinskaya oblast')

Improve the procedure for registering the staff. Fin. SSSR 19 no. 8:76
Ag '58. (MIRA 11:9)

1. Zaveduyushchiy Kopeyskim gorfinotdelom (for Khazanova); 2. Starshiy inspektor po shtatam Kopeyskogo gorfinotdela (for Platonova).
(Wages--Accounting)

PLATONOVA, O. P., Cand Bio Sci -- (diss) " *Platono*
of the lower Kama and middle Volga." Kazan' 1958,
17 pp. (Kazan' Order of Labor Red Banner State
Univ in F.I. Ul'yanova-Lenin) 150 copies (KL, 39-58, 100)

BEKKER, Z.E.; PLATONOVA, M.V.; SUPRUN, T.P.

Antagonistic fungi and soil formation. Izv. AN SSSR, Ser. Biol.
no. 5: 765-772 S-O '59. (MIRA 13:2)

1. The Faculty of Biology and Soil Sciences, the State Uni-
versity, Moscow.
(Soil micro-organisms) (Bacterial antagonism)
(Soil formation)

The Polarographic Reaction of Germanium on the Dropping Mercury Electrode 78-3-4-29/38

between the quantity of the diffusion current of Ge^{2+} and the concentration. The semiwave potential and the constant of the diffusion current were determined. The temperature coefficient of the diffusion current of bivalent germanium amounts to 0,5 % per $1^{\circ}C$. The magnitude of the temperature coefficient shows that in the reduction of bivalent germanium in hydrochloric acid a reversible process takes place. There are 8 figures, 6 tables and 8 references, 4 of which are Soviet.

ASSOCIATION: Krasnoznamennaya Voyennaya Inzhenernaya Akademiya svyazi imeni S. M. Budennogo
(Krasnoznamennaya Military Engineering Academy for Communications imeni S. M. Budenny)

SUBMITTED: May 15, 1957

Card 2/2

AUTHOR: Platonova, M. N.

78-3-4-29/38

TITLE: The Polarographic Reaction of Germanium on the Dropping-Mercury Electrode (Polarograficheskoye povedeniye germaniya na rtutnom kapel'nom elektrode)

PERIODICAL: Zhurnal Neorganicheskoy Khimii, 1958, Vol. 3, Nr 4, pp. 1002-1007 (USSR)

ABSTRACT: The polarographic reaction of tetravalent germanium in solutions of NH_4Cl , NH_4OH , lithium chloride and lithium hydroxide was investigated. A direct interdependence between the diffusion current of Ge^{4+} and the concentration was found. The semiwave potential and the constant of the diffusion current were determined. The reducibility of the tetravalent germanium ion in the above-mentioned solutions has irreversible character. The temperature coefficient of the diffusion current in lithium hydroxide amounts to 2 % per 1°C . The reaction of bivalent germanium in hydrochloric acid solution was also investigated polarographically. A direct interdependence exists

Card 1/2

PLATONOVA, M.N.

Polarographic behavior of germanium on a dropping mercury electrode.
Zhur.neorg.khim. 3 no.4:1002-1007 Ap '58. (MIRA 11:4)

1. Krasnoznamennaya Voyennaya inzhenernaya akademiya svyazi im. S.M.
Budennogo.

(Polarography) (Germanium)
(Electrodes, dropping mercury)

Investigation of the Submicroporosity of Low
absorbing Materials by the Method of Multiple
Small-angle X-Ray Scattering

81622
S/181/60/002/06/10/050
B122/B063

character, and the two curves run in parallel in their linear part. The radius R is determined from the slope of these straight lines ($R = 270 \text{ \AA}$). A maximum pore radius of 235 \AA results from the usual small-angle scattering from thin samples. The multiple X ray small angle scattering simplifies and facilitates the determination of the porosity of a material and is well applicable for the determination of the radii R of non-homogeneous regions larger than 100 \AA . Comparative data by S. M. Astrakhantses and Ya. S. Umanskiy (Ref. 8) (the former also supplied the beryllium samples) were also used to interpret the results (Table). There are 4 figures, 1 table, and 9 references: 2 Soviet, 2 American, 1 French.

ASSOCIATION: Institut mashinovedeniya AN SSSR, Moskva (Institute of
Machine Construction of the AS USSR, Moscow)

SUBMITTED: August 31, 1959

Case 3/3

...
 in... this is facilitated by the...
 Ca.../...

Investigation of the Submicroporosity of Low-
 absorbing Materials by the Method of Multiple
 Small-angle X-Ray Scattering

81522

S/181/60/062/06/10/050
 B122/B063

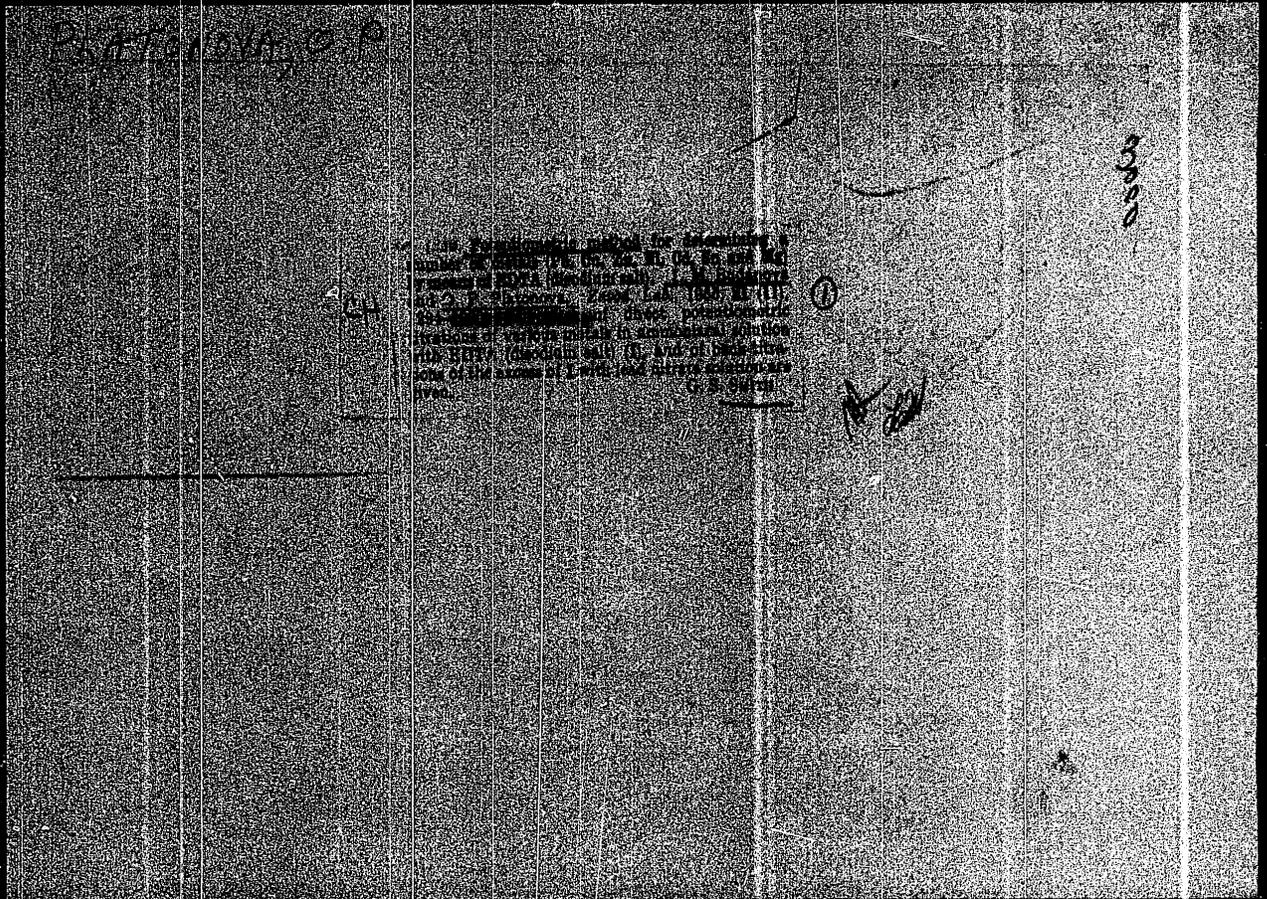
Again, high scattering power is secured by the use of materials with large R . However, a comparison between the results of the usual small-angle scattering and those by Lambert and Guinier is not so easy. In this connection, the paper under review reports on studies made on nonannealed, low-absorbing BeO , which exhibits a considerable inner submicroporosity (fluctuation of pore size from 20 to 250 Å). With a view to simplifying the determination of L , the authors developed a method allowing the direct determination of L from the measurement of the scattering intensities I_0 , I_1 from two different distances from the counter (one being in the immediate vicinity to the counter) (Figs. 1 and 2). The setup described was connected to a YPC-50W (URS-50I) apparatus. Scattering curves were repeatedly drawn of 11 samples with a thickness ranging from 0.06 to 0.80 g/cm², and the curves $(L_0/L_1)^2$ (integr) (1), L_0/L_1 (2), and I_0/I_1 (3) were drawn as functions of m (Fig. 4). In the case of a small m , curves 1 and 2 were found to differ from one another. In the case of larger m and prevailing of multiple scattering, the intensity distribution curve assumes a Gaussian

Card 2/3

PIATONOVA, G.D.

Characteristics of the feeding habits of beak-propagated insects
in the first years after the construction of Ingyskaya Reservoir.
Zool. zhur. 43 no. 5:706-712 '61

1. Kafedra zoologii pozvonochnykh zhivotnykh gosudarstvennogo
universiteta.



PLATONOVA, O.P.; ZAYTSEVA, G.M.

Uncompensated potentiometric method for determining copper with
the aid of salicylaldoxime. Zav.lab.22 no.2:165-166 F '56.
(MLRA 9:6)
(Oximes) (Copper--Analysis) (Potentiometric analysis)

PLATONOVA, O.P.

~~Potentiometric uncompensated method of copper determination with salicylaldehyde.~~ ~~O. P. Platono~~ ~~va and G. M. Zaitseva. Zavodskaya Lab. 4: 100-102 (1950).~~ Cu potentiometric titration with salicylaldehyde, with a bimetallic electrode pair and at a pH 4.5-6.5 is much quicker than any other highly accurate Cu method. Cu in steel and cast iron is detd. in a NaF soln. at the same pH, without filtering off the Fe fluoride. The pptn. of Fe is made necessary by the formation of complex Fe(III) salicylate in an AcOH soln.; 1-2% Fe does not interfere with the titration. Not over 15-20 min. time is required for a single Cu detn., and the time is reduced to 10 min. with a larger no. of detns. SO_4^{--} , Cl^- , NO_3^- , and CH_3COO^- do not affect the results.

W. M. Sternberg

3

2

Handwritten initials

Handwritten initials

PHATONOVA, OP.

2030. Potentiometric non-compensated method for determining copper by means of salicylaldoxime.

O. P. Platonova and G. M. Zaitseva. *Zavod. Lab.*

1960, 23 (2), 165-168.—To determine Cu in aluminum and magnesium alloys and bronzes, the sample (0.25 to 1 g) is dissolved in dil. HCl (1 + 1) [with the addition of a few drops of dil. HNO₃ (1 + 1)], dil. HNO₃ (1 + 1) or aqua regia. The solution is evaporated to a syrup, which is then dissolved in water. The soln., or a suitable aliquot portion (25 ml), is mixed with 15 ml of 20 per cent. ammonium acetate soln. and then with NaCl to saturation. The Cu is titrated potentiometrically with a standard soln. of salicylaldoxime, with tungsten-platinum electrodes. To determine Cu in plating baths, a sample containing 0.01 to 0.03 g of Cu is evaporated with HCl or H₂SO₄ to destroy cyanides, the residue is treated with ammonium acetate and NaCl as described above, and the soln. (120 ml) is titrated with 1 per cent. salicylaldoxime soln. The titrant is prepared by adding dropwise 95 ml of water at 30° C to a soln. of 1 g of salicylaldoxime in 5 ml of ethanol. There is no interference from SO₄²⁻, Cl⁻, NO₃⁻, Pb, metastannic acid and small amounts of Fe.

G. S. SMITH

Chem

3

5
0
0
0

Handwritten initials

BUDANOVA, L.M.; PLATONOVA, O.P.

Potentiometric method of determining a number of metals
(Pb,Cu,Zn,Ni,Cd,Sn,Mg) with the aid of trilon B. Zav. lab.
21 no.11:1294-1300 '55. (MLRA 9:2)
(Acetic acid) (Metals--Analysis)

EL'FOND, M.A., dotsent; PLATONOVA, N.P., vrach

Prevention of skin diseases in the workers of the "Oktabr"
Plant. Nauch. trudy Kub. gos. med. inst. 19:48-55 '62.

(MIRA 17:8)

1. Iz kafedry kozhnykh i venericheskikh bolezney (zaveduyushchiy
prof. L.A. Neradov) Kubanskogo gosudarstvennogo meditsinskogo
instituta.

SERDYUCHENKO, D.P.; PLATONOVA, N.Kh.; SMIRNOVA, N.V.; NESMEYANOV, A.N., akademik, glavnyy red.; TOPCHIYEV, A.V., akademik, zam.glavnogo red.; ISAKOVA, O.V., otv.red.; LIKHTENSHTEYN, Ye.S., otv.red.; SHUNKOV, V.I., otv.red.; MISHINA, R.L., red.izd-va; YEGOROVA, N.F., tekhn.red.

Petr Nikolaevich Chirvinskii, 1880-1955. Vstup.stat'ia D.P. Serdiuchenko i N.Kh.Platonova. Bibliografiia sost.N.V.Smirnovoi. Moskva, 1960. 93 p. (Materialy k biobibliografii uchenykh SSSR. Seriia geologicheskikh nauk, no.17). (MIRA 14:2)

1. Akademiya nauk SSSR.
(Bibliography--Chirvinskii, Petr Nikolaevich, 1880-1955)

ACC NR: AT7001812 SOURCE CODE: UR/2778/66/000/015/0066/0071

AUTHOR: Usoltsev, A. V.; Platonova, N. F.

ORG: none

TITLE: Compensation method of testing bimetallic heat sensors

SOURCE: Leningrad. Nauchno-issledovatel'skiy institut gidrometeorologicheskogo priborostroyeniya. Trudy, no. 15, 1966, 66-71

TOPIC TAGS: test method, compensation method, compensation test method, bimetallic sensor, bimetallic heat sensor, sensor, sensing element, measuring device, test instrumentation, meteorology

ABSTRACT: A method of testing bimetallic heat sensors used in meteorological instruments is discussed. Results obtained in testing the operational stability of sensors made from the heat-sensitive metals TB-3 and TB-72 are cited as examples. Orig. art. has: 2 figures and 1 table. [Translation of authors' abstract] [SP]

SUB CODE: 08, 09/SUBM DATE: none/OTH REF.: 001/

Card 1/1

AKRAMKHODZHAYEV, A.M.; AKHMEDZHANOV, M.A.; BABAYEV, A.G.; BABAYEV, K.L.;
BATALOV, A.B.; BASHAYEV, N.P.; BAYMUKHAMEDOV, Kh.N.; BRAGIN,
K.A.; BORISOV, O.M.; GABRIL'YAN, A.Sh.; GAR'KOVETS, V.G.;
GOR'KOVVOY, O.P.; GRIGORYANTS, S.V.; IBADULLAYEV, S.I.; ISMAILOV,
M.I.; ISAMUKHAMEDOV, I.M.; KAKHKHAROV, A.; KENESARIN, N.A.;
KRYLOV, M.M.; KUCHUKOVA, M.S.; LORDKIPANIDZE, L.N.; MAVLYANOV,
G.A.; MOTSOIKINA, T.M.; MALAKHOV, A.A.; MIRBABAYEV, M.Yu.;
MIRKHODZHIYEV, I.M.; MUSIN, R.A.; NABIYEV, K.A.; PETROV, N.P.;
POPOV, V.I.; PLATONOVA, N.A.; RYZHKOV, O.A.; SAYDALIYEVA, M.S.;
SERGUN'KOVA, O.I.; SLYADNEV, A.F.; TULYAGANOV, Kh.T.; UKLONSKIY,
A.S.; KHAMRABAYEV, I.Kh.; KHODZHIBAYEV, N.N.; CHUMAKOV, I.D.;
SHAVLO, S.G.

Khabib Mukhamedovich Abdullaev; obituary. Uzb.geol.zhur. 6
no.4:7-9 '62. (MIRA 15:9)
(Abdullaev, Khabib Mukhamedovich, 1912-1962)

PLATONOVA, N.F.

A ventilated shelter. Trudy NITOP no.12428-52 168.

(MIRA 18-11)

PLATONOVA, M.N.

Current efficiency in the anodic dissolution of germanium.
Zhur.prikl.khim. 35 no.2:334-339 F 162. (GRA 15:2)
(Germanium--Electric properties)

PLATONOVA, M.N.

Polarographic determination of acrylonitrile in alcoholic aqueous solutions. Zhur.anal.khim. 11 no.3:310-312 My-Je '56. (MIRA 9:8)

1. Inzhenernaya akademiya svyazi imeni S.M. Budennogo.
(Acrylonitrile) (Polarography)

PLATONOVA, M.N.

Using the polarographic method for the control of polymerization
processes. Zav. Lab. 23 no. 5: 539-540 '57. (MLRA 10:8)
(Polarography) (Polymerization)

PLATONOVA, M.H.

Polarographic determination of germanium in solutions
after etching of its surface. Zav.kav. 36 no.7:795-796
'60. (MIRA 13:7)

(Germanium--Analysis)

PASECHNIK, M.S., doktor tekhn. nauk; ZHEL'VIS, A.I. kand. tekhn. nauk; KORBUT, V.A.; PLATONOVA, M.N.; SEMYALINA, T.S.; TSINTSIUS, V.M.; STRELE, L.A., red.

[Manual on general chemistry and physicochemical methods of analysis] Uchebnoe posobie po obshchei khimii i fiziko-khimicheskim metodam analiza. [by] M.S.Pasechnik i dr. pod obshchei red. M.S.Pasechnika i A.I.Zhel'vis (chast' 1). Leningrad, 1965. 204 p. (MIRA 19:1)

1. Leningrad. Lesotekhnicheskaya akademiya.

1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 48 49 50

A B C D E F G H I J K L M N O P Q R S T U V W X Y Z AA AB AC AD AE AF AG AH AI AJ AK AL AM AN AO AP AQ AR AS AT AU AV AW AX AY AZ BA BB BC BD BE BF BG BH BI BJ BK BL BM BN BO BP BQ BR BS BT BU BV BW BX BY BZ CA CB CC CD CE CF CG CH CI CJ CK CL CM CN CO CP CQ CR CS CT CU CV CW CX CY CZ DA DB DC DD DE DF DG DH DI DJ DK DL DM DN DO DP DQ DR DS DT DU DV DW DX DY DZ EA EB EC ED EE EF EG EH EI EJ EK EL EM EN EO EP EQ ER ES ET EU EV EW EX EY EZ FA FB FC FD FE FF FG FH FI FJ FK FL FM FN FO FP FQ FR FS FT FU FV FW FX FY FZ GA GB GC GD GE GF GG GH GI GJ GK GL GM GN GO GP GQ GR GS GT GU GV GW GX GY GZ HA HB HC HD HE HF HG HH HI HJ HK HL HM HN HO HP HQ HR HS HT HU HV HW HX HY HZ IA IB IC ID IE IF IG IH II IJ IK IL IM IN IO IP IQ IR IS IT IU IV IW IX IY IZ JA JB JC JD JE JF JG JH JI JJ JK JL JM JN JO JP JQ JR JS JT JU JV JW JX JY JZ KA KB KC KD KE KF KG KH KI KJ KL KM KN KO KP KQ KR KS KT KU KV KW KX KY KZ LA LB LC LD LE LF LG LH LI LJ LK LL LM LN LO LP LQ LR LS LT LU LV LW LX LY LZ MA MB MC MD ME MF MG MH MI MJ MK ML MN MO MP MQ MR MS MT MU MV MW MX MY MZ NA NB NC ND NE NF NG NH NI NJ NK NL NO NP NQ NR NS NT NU NV NW NX NY NZ OA OB OC OD OE OF OG OH OI OJ OK OL OM ON OP OQ OR OS OT OU OV OW OX OY OZ PA PB PC PD PE PF PG PH PI PJ PK PL PM PN PO PP PQ PR PS PT PU PV PW PX PY PZ QA QB QC QD QE QF QG QH QI QJ QK QL QM QN QO QP QQ QR QS QT QU QV QW QX QY QZ RA RB RC RD RE RF RG RH RI RJ RK RL RM RN RO RP RQ RR RS RT RU RV RW RX RY RZ SA SB SC SD SE SF SG SH SI SJ SK SL SM SN SO SP SQ SR SS ST SU SV SW SX SY SZ TA TB TC TD TE TF TG TH TI TJ TK TL TM TN TO TP TQ TR TS TT TU TV TW TX TY TZ UA UB UC UD UE UF UG UH UI UJ UK UL UM UN UO UP UQ UR US UT UU UV UW UX UY UZ VA VB VC VD VE VF VG VH VI VJ VK VL VM VN VO VP VQ VR VS VT VU VW VX VY VZ WA WB WC WD WE WF WG WH WI WJ WK WL WM WN WO WP WQ WR WS WT WU WV WW WX WY WZ XA XB XC XD XE XF XG XH XI XJ XK XL XM XN XO XP XQ XR XS XT XU XV XW XX XY XZ YA YB YC YD YE YF YG YH YI YJ YK YL YM YN YO YP YQ YR YS YT YU YV YW YX YY YZ ZA ZB ZC ZD ZE ZF ZG ZH ZI ZJ ZK ZL ZM ZN ZO ZP ZQ ZR ZS ZT ZU ZV ZW ZX ZY ZZ

INDEX AND SUBJECTS

INDEXES AND SUBJECTS

Oxidation of chromium hydroxide and chromite by the air in an alkaline medium.
V. V. IPAT'EV, JR. AND M. N. PLATONOVA. *J. Applied Chem (U.S.S.R.)* 4, 683 (1931)

(1931) -- Cr^{III} is oxidized rapidly to Cr^{VI} in an alk. medium by the air under pressure. The oxidation also takes place in K₂CrO₄ soln. but at a lower rate. In the absence of O₂ the Cr(OH)₃ is not oxidized by H₂O at 300°. In the presence of O₂ partial oxidation takes place but is soon interrupted due to the accumulation of H₂O₂. Increased concn of OH⁻ hastens the reaction. Oxidation of Cr in chromite is considerably less rapid. Elevated temps. and an excess of OH⁻ speed the reaction. No oxidation takes place in K₂CrO₄ soln.

V. KALICHVANSKY

AS A S L A METALLURGICAL LITERATURE CLASSIFICATION

432

SD 11 15 AT NO 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 48 49 50

1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 48 49 50 51 52 53 54 55 56 57 58 59 60 61 62 63 64 65 66 67 68 69 70 71 72 73 74 75 76 77 78 79 80 81 82 83 84 85 86 87 88 89 90 91 92 93 94 95 96 97 98 99 100

A B C D E F G H I J K L M N O P Q R S T U V W X Y Z AA BB CC DD EE FF GG HH II JJ KK LL MM NN OO PP QQ RR SS TT UU VV WW XX YY ZZ

SEPARATION OF TRACES OF IRON FROM ALUMINUM CHLORIDE. V. V. IPAT'EV, JR., AND M. N. PLATONOVA. *J. Applied Chem. (U.S.S.R.)* 4, 701-3(1931). Fe can be sep'd from AlCl₃ solns. by heating with HCl under pressure. Elevated temp. speeds up the reaction. Expts. were made at 300° and with 1.4 N HCl, at which concn. FeCl₃ does not hydrolyze, while AlCl₃ hydrolyzes in solns. weaker than 2.8 N in HCl. The method presents considerable difficulties for com. application. V. KALICHEVSKY

ASME-ISA METALLURGICAL LITERATURE CLASSIFICATION

1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 48 49 50 51 52 53 54 55 56 57 58 59 60 61 62 63 64 65 66 67 68 69 70 71 72 73 74 75 76 77 78 79 80 81 82 83 84 85 86 87 88 89 90 91 92 93 94 95 96 97 98 99 100

1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 48 49 50 51 52 53 54 55 56 57 58 59 60 61 62 63 64 65 66 67 68 69 70 71 72 73 74 75 76 77 78 79 80 81 82 83 84 85 86 87 88 89 90 91 92 93 94 95 96 97 98 99 100

U I J K L M N O P Q R S T U V W X Y Z AA AB AC AD AE AF AG AH AI AJ AK AL AM AN AO AP AQ AR AS AT AU AV AW AX AY AZ BA BB BC BD BE BF BG BH BI BJ BK BL BM BN BO BP BQ BR BS BT BU BV BW BX BY BZ CA CB CC CD CE CF CG CH CI CJ CK CL CM CN CO CP CQ CR CS CT CU CV CW CX CY CZ DA DB DC DD DE DF DG DH DI DJ DK DL DM DN DO DP DQ DR DS DT DU DV DW DX DY DZ EA EB EC ED EE EF EG EH EI EJ EK EL EM EN EO EP EQ ER ES ET EU EV EW EX EY EZ FA FB FC FD FE FF FG FH FI FJ FK FL FM FN FO FP FQ FR FS FT FU FV FW FX FY FZ GA GB GC GD GE GF GG GH GI GJ GK GL GM GN GO GP GQ GR GS GT GU GV GW GX GY GZ HA HB HC HD HE HF HG HH HI HJ HK HL HM HN HO HP HQ HR HS HT HU HV HW HX HY HZ IA IB IC ID IE IF IG IH II IJ IK IL IM IN IO IP IQ IR IS IT IU IV IW IX IY IZ JA JB JC JD JE JF JG JH JI JJ JK JL JM JN JO JP JQ JR JS JT JU JV JW JX JY JZ KA KB KC KD KE KF KG KH KI KJ KL KM KN KO KP KQ KR KS KT KU KV KW KX KY KZ LA LB LC LD LE LF LG LH LI LJ LM LN LO LP LQ LR LS LT LU LV LW LX LY LZ MA MB MC MD ME MF MG MH MI MJ MK ML MN MO MP MQ MR MS MT MU MV MW MX MY MZ NA NB NC ND NE NF NG NH NI NJ NK NL NO NP NQ NR NS NT NU NV NW NX NY NZ OA OB OC OD OE OF OG OH OI OJ OK OL OM ON OP OQ OR OS OT OU OV OW OX OY OZ PA PB PC PD PE PF PG PH PI PJ PK PL PM PN PO PP PQ PR PS PT PU PV PW PX PY PZ QA QB QC QD QE QF QG QH QI QJ QK QL QM QN QO QP QQ QR QS QT QU QV QW QX QY QZ RA RB RC RD RE RF RG RH RI RJ RK RL RM RN RO RP RQ RR RS RT RU RV RW RX RY RZ SA SB SC SD SE SF SG SH SI SJ SK SL SM SN SO SP SQ SR SS ST SU SV SW SX SY SZ TA TB TC TD TE TF TG TH TI TJ TK TL TM TN TO TP TQ TR TS TT TU TV TW TX TY TZ UA UB UC UD UE UF UG UH UI UJ UK UL UM UN UO UP UQ UR US UT UU UV UW UX UY UZ VA VB VC VD VE VF VG VH VI VJ VK VL VM VN VO VP VQ VR VS VT VU VV VW VX VY VZ WA WB WC WD WE WF WG WH WI WJ WK WL WM WN WO WP WQ WR WS WT WU WV WW WX WY WZ XA XB XC XD XE XF XG XH XI XJ XK XL XM XN XO XP XQ XR XS XT XU XV XW XX XY XZ YA YB YC YD YE YF YG YH YI YJ YK YL YM YN YO YP YQ YR YS YT YU YV YW YX YY YZ ZA ZB ZC ZD ZE ZF ZG ZH ZI ZJ ZK ZL ZM ZN ZO ZP ZQ ZR ZS ZT ZU ZV ZW ZX ZY ZZ

COMMON ELEMENTS

CHLORINATION AS A METHOD FOR THE PURIFICATION OF WASTE WATERS. V. E. Kuni and M. N. Platonova. *Vostochno-Zhena Smit. Tekh.* 1937, No. 9, 61-9; *Khim. Referat. Zhur.* 1938, No. 6, 81. In the purification of waste waters of textile mills, doses of 395-450 mg./l. caused a rapid coagulation and pptn. of all suspended and of a part of the dissolved org. substances. The physical properties of water improved after its filtration through paper or sand. Approx. 30% of Cl remained unused. As a result of chlorination the amt. of the pptd. substances reached 17.5 mg./l.; approx. 1/3 of the org. substances contained in water were removed. W. R. Hein

AYW 51A METALLURGICAL LITERATURE CLASSIFICATION

11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 48 49 50 51 52 53 54 55 56 57 58 59 60 61 62 63 64 65 66 67 68 69 70 71 72 73 74 75 76 77 78 79 80 81 82 83 84 85 86 87 88 89 90 91 92 93 94 95 96 97 98 99 100

Current yield for ...

S/080/52/035/002/012/022
D244/D302

J. Electroch. Soc., 103, 252, 1956; W. Pugh, J. Chem. Soc., 11,
1539, 1929.

SUBMITTED: January 25, 1961

Card 3/3

X

Current yield for ...

3/030/02/035/052/0'2,022
D244/D302

graphically in the form of Ge^{4+} . The formation of deposit on Ge-electrodes took place at the current densities of 5 and 8 mA/cm^2 . It was shown that Ge dissolves in the form of Ge^{4+} the yield with respect to current passed being near to 100%. This yield was observed at 1 and 2.5 mA/cm^2 in the absence of visible deposit on the electrode. The yield also was near to 100% when the deposit formation took place, if Ge in solution was added to that in the deposit. For the high current densities the yield was considerably higher than 100%. This was due to the interaction of Ge^{4+} with the surface of unoxidized Ge with the formation of Ge^{2+} . When the deposit dissolved, the unstable GeO was transferred to GeO_2 which led to the total quantity of Ge being greater than 100%. It was shown that the deposits on p-Ge appear at potentials which are more positive than those for n-Ge. There are 2 figures, 2 tables and 9 references: 4 Soviet-bloc and 5 non-Soviet-bloc. The references to the English-language publications read as follows: P. F. Schmidt, M. Bromgren, J. Electroch. Soc., 106, 8, 694, 1959; D. R. Turner,

Card 2/3